

Insertion of Terminal Phosphinidene Complexes into a Nickel–Carbon Bond; X-Ray Crystal Structure of the Complex $[\text{CpNi}\{\eta\text{-Me-P-CPh=CPh-NiCp}\}\text{P-W}(\text{CO})_5]$ (Cp = Cyclopentadienyl)

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New phosphametallacyclobutene complexes (**4a–c**) have been obtained by the reactions of (7-methyl-7-phosphanorbornadiene)pentacarbonyltungsten (**1**) with some alkynebis(cyclopentadienylnickel) complexes.

In a series of recent papers, we have described the reactivity of terminal phosphinidene complexes $[\text{RP}=\text{W}(\text{CO})_5]$ ¹ towards common organic compounds. Addition of the phosphinidene complexes to olefins² and alkynes,¹ and insertion reactions into N–H, O–H,³ and strained N–C and O–C⁴ bonds were

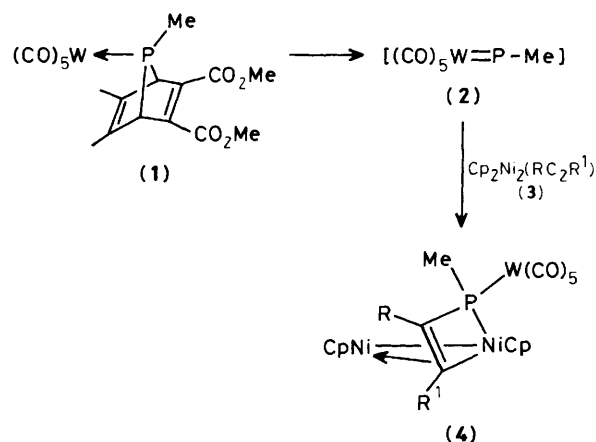
observed. However, the behaviour of such phosphinidene complexes towards organometallic species is still almost unexplored: only the reactions between phosphinidenes and Fischer-type carbene complexes have been studied.⁵ We report here a new reaction of terminal phosphinidene com-

plexes (2) with $\text{Cp}_2\text{Ni}_2(\text{C}_2\text{R}_2)$ clusters. The transient species (2) is generated, as usual, from the corresponding 7-phosphanorbornadiene- $\text{W}(\text{CO})_5$ complex (1) by thermolysis at 120°C (method A) or at 60°C with CuCl as catalyst (method B). It reacts with the bimetallic nickel alkyne complexes (3) to give the new phosphametallacyclobutene complexes (4).

According to the ^{31}P n.m.r. spectra of the crude reaction mixtures, the complexes (4a-c) are the only products formed. Purifications were performed by column chromatography on silica gel and crystallization. The phosphinidene complex (2) inserts selectively into the $\text{Ni}-\text{C}(\text{CO}_2\text{Et})$ bond of (3b) and preferentially into the $\text{Ni}-\text{CH}$ bond of (3c).

The structural formulation of the complexes (4a-c) was based on ^{31}P , ^{13}C , and ^1H n.m.r., i.r., and mass spectroscopic data† and confirmed by an X-ray diffraction study of (4a).‡

The adducts (4a-c) contain the $[\text{MeP}\rightarrow\text{W}(\text{CO})_5]$ entity inserted into a $\text{Ni}-\text{C}$ bond of the starting alkynebis(cyclopentadienylnickel) cluster. The co-ordination mode of the second nickel atom is unchanged. Similar phosphametallacyclobutene complexes are obtained otherwise by photochemical insertion of some alkynes into the $\text{P}-\text{Fe}$ bond of the trimetallic phosphinidene cluster $(\mu_3\text{-PR})\text{Fe}_3(\text{CO})_{10}$.⁶ The insertion of



	R	R ¹	Method	Yield (%)	$\delta(^{31}\text{P})$ (J_{PW}/Hz)
a;	Ph	Ph	B	40	-113.4 (222)
b;	CO_2Et	Ph	A	42	-99.8 (229)
c;	H	Ph	B	25	-119.9 (217)
	Ph	H			-139.7

† (4a): ^1H n.m.r. (C_6D_6) δ 1.73 [d, $^2J(\text{H}-\text{P})$ 5.1 Hz, Me], 4.92 (s, Cp), and 5.07 (s, Cp); ^{13}C n.m.r. (CDCl_3) δ 31.96 [d, $^1J(\text{C}-\text{P})$ 8.0 Hz, Me], 70.69 [d, $^1J(\text{C}-\text{P})$ 13.6 Hz, P-C-Ph], 92.09 (Cp), 92.91 (Cp), 139.7 (d, 2J 7.5 Hz), 147.50, 153.89, 198.80 [d, $^2J(\text{C}-\text{P})$ 6.5 Hz, *cis*-CO], and 200.55 [d, $^2J(\text{C}-\text{P})$ 23.1 Hz, *trans*-CO]; i.r. (decalin) $\nu(\text{CO})$ 2070m, 1935s, and 1930vs cm^{-1} ; mass spectrum (70 eV, ^{58}Ni , ^{184}W) m/z 766 ($M-\text{CO}$, 8%), 710 ($M-3\text{CO}$, 21%), 654 ($M-5\text{CO}$, 13%), and 476 ($M-5\text{CO}-\text{C}_2\text{Ph}_2$, 100%).

(4b): ^1H n.m.r. (C_6D_6) δ 1.02 [t, $^3J(\text{H}-\text{H})$ 7.0 Hz], 1.72 [d, $^2J(\text{H}-\text{P})$ 5.6 Hz, PMe], 3.9 (m, CH_2), 4.86 (s, Cp), and 5.24 (s, Cp); ^{13}C n.m.r. (CDCl_3) δ 14.18 (CH_2CH_3), 31.31 [d, $^1J(\text{C}-\text{P})$ 6.0 Hz, PMe], 57.03 [d, $^1J(\text{C}-\text{P})$ 13.7 Hz, P-C- CO_2], 60.72 (CH_2), 91.99 (Cp), 93.03 (Cp), 128.33, 128.53, 129.18, 146.89, 160.13, 167.62 [d, $^2J(\text{C}-\text{P})$ 7.5 Hz, CO_2Et], 198.69 [d, $^2J(\text{C}-\text{P})$ 6.5 Hz, *cis*-CO], and 200.91 [d, $^2J(\text{C}-\text{P})$ 25.7 Hz, *trans*-CO]; i.r. (decalin) $\nu(\text{CO})$ 2060m, 1940sh, and 1930vs cm^{-1} ; $\nu(\text{ester C-O})$ 1700 cm^{-1} ; mass spectrum (70 eV, ^{58}Ni , ^{184}W) m/z 762 ($M-\text{CO}$, 13%), 706 ($M-3\text{CO}$, 30%), 650 ($M-5\text{CO}$, 10%), and 476 ($M-5\text{CO}-\text{PhC}_2\text{CO}_2\text{Et}$, 100%).

(4c): mixture of two isomers, the major characterised as follows: ^1H n.m.r. (C_6D_6) δ 1.52 [d, $^2J(\text{H}-\text{P})$ 5.4 Hz, Me], 3.74 [d, $^2J(\text{H}-\text{P})$ 19.3 Hz, P-CH], 4.92 (s, Cp), and 5.09 (s, Cp); ^{13}C n.m.r. (CDCl_3) δ 34.43 (PMe), 52.13 [d, $^1J(\text{C}-\text{P})$ 14.6 Hz, PCH], 91.01 (Cp), 92.27 (Cp), 128.60, 147.99, 159.51, and 198.86 [d, $^2J(\text{C}-\text{P})$ 6.0 Hz, *cis* CO]; Mass spectrum (70 eV, ^{58}Ni , ^{184}W) m/z 690 ($M-\text{CO}$, 6%), 662 ($M-2\text{CO}$, 17%), 634 ($M-3\text{CO}$, 38%), 578 ($M-5\text{CO}$, 32%), and 476 ($M-5\text{CO}-\text{PhC}_2\text{H}$, 100%).

‡ Crystal data for (4a): $\text{C}_{30}\text{H}_{25}\text{O}_5\text{PNi}_2\text{W}$, $M = 795.76$, orthorhombic, space group $P2_12_1$ (no. 19), $a = 11.888(1)$, $b = 14.479(1)$, $c = 17.000(2)$ Å, $U = 2926.2(8)$ Å³, $Z = 4$, $D_c = 1.806$ g cm^{-3} , $\text{Mo-K}\alpha$ (0.71073 Å) radiation, $\mu = 53.8$ cm^{-1} , $F(000) = 1552$. Intensity data were collected with an Enraf-Nonius CAD4 diffractometer in the $\Theta-2\Theta$ scan mode. The intensities of 4745 reflections were measured in the range $2^\circ \leq \Theta \leq 30^\circ$.

The structure was solved by Patterson and Fourier methods and refined by full-matrix least-squares using the 2837 unique reflections having $F^2 > 3\sigma(F^2)$. The final $R_1(F)$ and $R_2(F)$ values are 0.032 and 0.034, respectively. The goodness of fit is 1.08. The enantiomeric structure yields $R = 0.054$. Atomic co-ordinates, bond and angle tables, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.

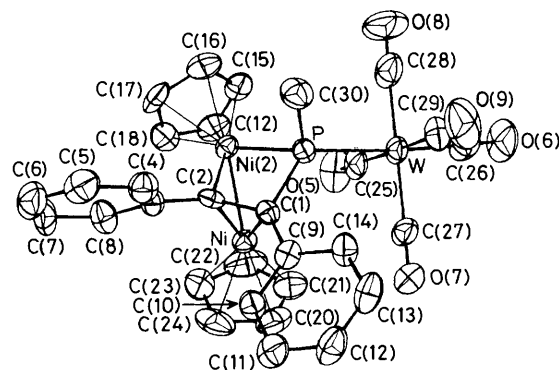


Figure 1. X-Ray crystal structure of $[\text{CpNiW}(\text{CO})_5\{\text{MeP}-\text{CPh}=\text{CPh}-\text{NiCp}\}]$ (4a). Principal bond distances (Å): Ni(2)-P 2.197(2), P-C(1) 1.812(8), C(1)-C(2) 1.41(1), C(2)-Ni(2) 1.920(7), P-C(30) 1.843(9), P-W 2.532(2), Ni-Ni(2) 2.345(1), Ni-C(2) 1.876(8), Ni-C(1) 2.006(7). Bond angles ($^\circ$) Ni(2)-P-C(1) 81.0(3), P-C(1)-C(2) 98.5(5), C(1)-C(2)-Ni(2) 102.5(5), C(2)-Ni(2)-P 72.9(2), Ni(2)-P-W 122.88(9), Ni(2)-P-C(30) 104.7(4), C(2)-Ni-C(1) 42.5(3), C(2)-Ni(2)-Ni 51.0(2).

ML_2 ($M = \text{Ni}$, Pd , or Pt ; $L =$ phosphine) into phosphirenes yields similar heterocycles.⁷

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